

ALKALOIDS OF *ERYTHROXYLUM ZAMBESIACUM* ROOT-BARK*

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Key Word Index—*Erythroxylum zambesiaceum*; Erythroxylaceae; root-bark; tropane alkaloids; esterifying acids; chemotaxonomy

Abstract—Six new alkaloids characterized from the root-bark of *Erythroxylum zambesiaceum* were: 3 α -(3,4,5-trimethoxybenzoyloxy)nortropane, 3 α -(3,4,5-trimethoxybenzoyloxy)tropan-6 β -ol, 3 α -(3,4,5-trimethoxybenzoyloxy)-nortropan-6 β -ol, 6 β -benzoyloxytropan-3 α ,7 β -diol, 6 β -benzoyloxy-3 α -(3,4,5-trimethoxycinnamoyloxy)tropan-7 β -ol, and 7 β -acetoxo-6 β -benzoyloxy-3 α -(3,4,5-trimethoxycinnamoyloxy)tropane. Other bases identified included 3 α -(3,4,5-trimethoxybenzoyloxy)tropane, 3 α -(3,4,5-trimethoxycinnamoyloxy)tropane, 3 α -phenylacetoxytropan-6 β -ol, 3 α -(3,4,5-trimethoxybenzoyloxy)tropan-6 β ,7 β -diol, 6 β -benzoyloxytropan-3 α -ol, and 6 β -benzoyloxy-3 α -(3,4,5-trimethoxycinnamoyloxy)tropane; other bases could not be fully characterized. The chemotaxonomic implications of the esterifying acids are discussed.

INTRODUCTION

Erythroxylum zambesiaceum N. Robson, first described in 1962 [1], is included in *Flora Zambesiaca* (1963) [2] and *Flora of Tropical East Africa* (1984) in which it is listed incorrectly as *E. zambesianum* [3]. It is a shrub or small tree mainly confined to the Zambesi valley from Kariba to above the Victoria Falls and is related to the West African *E. mannii* Oliv. [1]. No mention of local uses or reports concerning the phytochemistry of *E. zambesiaceum* are to be found in the literature. As part of a wider investigation of the genus for tropane and related alkaloids having pharmacological and chemotaxonomic interest, we record here our findings on the basic constituents of the root-bark of this species.

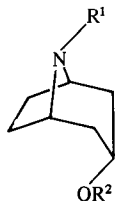
RESULTS AND DISCUSSION

Preliminary thin-layer chromatography (TLC) examination of the extracts of small quantities of root- and stem-bark showed an apparent similarity in the alkaloid spectrum of both sources; the root-bark was studied in detail and five principal bases and several minor ones were apparent, representing 0.28% of the dried plant material. In typical experiments 100–200 g of root-bark were exhaustively extracted with ether and the alkaloid extract fractionated by column partition chromatography at pH 6.8 with petrol (bp 60–80°), ether and chloroform in succession as eluants. Preparative TLC and the fractional crystallization of picrates gave further separations. Thirteen basic fractions were recognised and identified 1–13 in order of their ascending R_f values on TLC. For the characterization of the isolated tropane alkaloids the spectroscopic criteria previously outlined [4] and standard chemical procedures were employed. Fraction 1, the

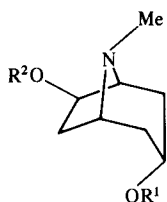
slowest running by TLC, was eluted from the partition column by chloroform and contained one principal and seven minor bases. Preparative TLC gave the major alkaloid which by IR and mass spectroscopy was shown to be an ester of a nortropanol and trimethoxybenzoic acid. The structure of the new alkaloid was confirmed as 3 α -(3,4,5-trimethoxybenzoyloxy)nortropane (**1a**) by comparison of the natural alkaloid picrate with that of the picrate of a synthetic sample of the base.

Fraction 2 (R_f 0.34, System B see Experimental) was not isolated in sufficient quantity for further investigation; fraction 3 was identified as 3 α -(3,4,5-trimethoxybenzoyloxy)tropane (**1b**), a base previously isolated [5] from *E. monogynum*, and fraction 4 as the analogous 3,4,5-trimethoxycinnamate (**1c**), a constituent of both *E. monogynum* root-bark [5] and *E. ellipticum* leaves [6]. A dextrorotatory base from fraction 5 had R_f 0.41 (System C see Experimental) and possessed ester and hydroxyl functions. The molecular formula, $C_{18}H_{25}NO_6$, and mass fragmentation pattern, $[M]^+$ and $[M-44]^+$ $[M-(CHOH-CH_2)]^+$, corresponded to tropane-3,6-diol esterified at C-3 with an acid, $C_{10}H_{12}O_5$, shown after hydrolysis and isolation to be 3,4,5-trimethoxybenzoic acid. Further support for the C-3 ester linkage was given by the mass spectrum of this base, as its 6 β -phenylacetyl derivative, $([M]^+$ and $[M-(PhCH_2OCOCH-CH_2)]^+)$. 1H NMR spectroscopy indicated 3 β - and 6 α -protons. The structure of the new alkaloid is therefore (+)-3 α -(3,4,5-trimethoxybenzoyloxy)tropan-6 β -ol (**2a**); the picrate has mp 214°. From the picrate mother-liquors of **2a**, fraction 6 was deposited as a gummy picrate from which a laevorotatory base was released; this base was characterized as 3 α -phenylacetoxytropan-6 β -ol (**2b**), a known constituent of *E. hypericifolium* root-bark [4]. A comparison of the 6 β -phenylacetyl derivative of this base with semi-synthetic compounds prepared from the (+)-, (–)- and (±)-tropanediols indicated the isolated alkaloid in this instance was an unequal mixture of the (+)- and (–)-esters; in contrast the alkaloid isolated from *E. hyper-*

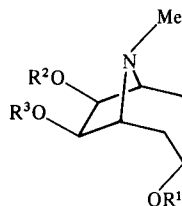
* Part 7 in the series 'Alkaloids of the Genus *Erythroxylum*'. For part 6 see ref. [10].



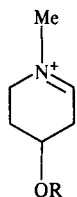
- 1a** $R^1 = \text{H}, R^2 = (\text{MeO})_3\text{C}_6\text{H}_2\text{CO (Tmb)}$
1b $R^1 = \text{Me}, R^2 = \text{Tmb}$
1c $R^1 = \text{Me}, R^2 = (\text{MeO})_3\text{C}_6\text{H}_2\text{CH:CHCO (Tmc)}$



- 2a** $R^1 = \text{Tmb}, R^2 = \text{H}$
2b $R^1 = \text{PhCH}_2\text{CO}, R^2 = \text{H}$
2c $R^1 = \text{H}, R^2 = \text{PhCO (Bz)}$
2d $R^1 = \text{Tmc}, R^2 = \text{Bz}$
2e $R^1 = R^2 = \text{PhCH}_2\text{CO}$



- 3a** $R^1 = \text{Tmb}, R^2 = R^3 = \text{H}$
3b $R^1 = R^3 = \text{H}, R^2 = \text{Bz}$
3c $R^1 = \text{Tmc}, R^2 = \text{Bz}, R^3 = \text{H}$
3d $R^1 = \text{Tmc}, R^2 = \text{Bz}, R^3 = \text{MeCO}$



	R	m/z
4a	Tmb	307
4b	H	113
4c	Bz	217
4d	Tmc	333
4e	PhCH ₂ CO	231

icifolium appeared to be derived exclusively from the (+)-diol.

Fraction 7 contained the principal alkaloid of the chloroform eluate and it was identified as 3 α -(3,4,5-trimethoxybenzoyloxy)tropan-6 β ,7 β -diol (**3a**) a known constituent of *E. monogynum* root bark [5]. Fraction 8, closely associated with fraction 7 in the chromatographic separations, had R_f 0.62 (System A see Experimental) and readily formed a picrate. Elemental analysis and spectroscopy indicated a molecular formula of $\text{C}_{15}\text{H}_{19}\text{NO}_4$ for the base. The mass spectrum of the picrate was characteristic of a tropanol esterified with benzoic acid and the presence of a prominent ion **4b** and the absence of an ion (**4c**) $[\text{M} - 60]^+$ indicated esterification at C-6 and a free hydroxyl at C-3.¹H NMR spectroscopy demonstrated 3 β -, 6 α - and 7 α -protons together with aromatic, *N*-methyl

and hydroxyl functions. The above data established the structure of the new alkaloid as 6 β -benzoyloxytropan-3 α ,7 β -diol (**3b**). The isomeric 3 α -benzoyl ester was reported [7] as a constituent of *E. australe* roots.

Fraction 9 constituted the later portion of the ether eluate of the partition column and contained a base having the spectroscopic characters of a synthetic sample of (\pm)-6 β -benzoyloxytropan-3 α -ol (**2c**). The natural alkaloid was dextrorotatory and appears to be identical with the (+)-alkaloid isolated by Lounasmaa *et al.* [8] from *Knightia strobilina*, (Proteaceae). From fraction 10, the main ether eluate of the column, a base of R_f 0.25 (System B see Experimental) was isolated, and purified by picrate formation. Elemental analysis and spectroscopy indicated that the base had the molecular formula $\text{C}_{17}\text{H}_{23}\text{NO}_6$, with ester and hydroxyl functions. Mass

spectrometry suggested a substituted nortropane-3,6-diol associated with a trimethoxybenzoic acid moiety; the 3,4,5-substitution pattern for the methoxy groups was established by ^1H NMR spectroscopy (singlet at δ 7.28), other isomers have dissimilar signals [9]. The distorted multiplet at δ 5.0, assigned to the 3β -proton, is similar in multiplicity and chemical shift to the 3β -proton of noratropine. Although m/z 293 $[\text{M} - (\text{C}(6)\text{HOH} - \text{C}(7)\text{H}_2)]^+$ was absent in the mass spectrum, the occurrence of the 6α -proton as a doublet of doublets at δ 4.56 confirms this substituent as the free alcohol rather than as an ester. Thus, this fraction was identified as 3α -(3,4,5-trimethoxybenzoyloxy)nortropane-6 β -ol. Similar bases esterified at C-3 and involving benzoic or phenylacetic acid have been characterized [10, 11] from the leaves of *E. macrocarpum* and *E. sideroxyloides* and from the roots of *E. cumense*.

The initial elution of the partition column with petrol gave three distinct fractions. Fraction 11 had R_f 0.70 (System B see Experimental) and was purified by picrate formation. Mass spectrometry gave M_r of 497 with a characteristic tropane-diol or -triol fragmentation. Esterification involving two different acids was indicated by the IR spectrum and by the $[\text{M}]^+$ ions of benzoic and trimethoxycinnamic acids and their acylium ions in the mass spectrum; attachment of the latter acid at C-3 was established by a low intensity ion at m/z 333 (**4d**). A free hydroxyl group was apparent in the IR spectrum. The ^1H NMR spectrum supported the mass spectral findings and, by the established criteria, showed the 3α -configuration of the ester group, the 3,4,5-substitution of the aromatic ring and the $6\beta, 7\beta$ -orientation of the remaining groups. The new alkaloid is therefore 6β -benzoyloxy- 3α -(3,4,5-trimethoxy-cinnamoyloxy)tropane-7 β -ol (**3c**); it had previously been suspected, but not confirmed, as a component of *E. monogynum* root-bark [12].

Fraction 12 consisted principally of a base R_f 0.70 (System D) which was purified by picrate formation. Elemental analysis and mass spectroscopy gave the molecular formula of the base as $\text{C}_{29}\text{H}_{33}\text{NO}_9$ and the IR spectrum showed three different carbonyl functions. In the mass spectrum a trisubstituted tropane was indicated and benzoic and trimethoxycinnamic acids and their acylium ions were detected. The trimethoxycinnamic acid formed an ester function at C-3 of the tropanol [m/z 333 (**4d**)], with confirmation following from the absence of ions m/z 217 [3-benzoate (**4c**)] and 155 (3-acetate). The acetate moiety was supported by the ion m/z 43 and confirmed by the ^1H NMR signal at δ 1.87, a three-proton singlet ascribable to the acetoxy group. The 3,4,5-trimethoxy substitution of the acid moiety and the $3\alpha, 6\beta, 7\beta$ -orientation of the tropane substituents followed from the chemical shifts for the aromatic singlet and the CH-OH protons at δ 6.88, 5.77 and 5.92, respectively. The remaining signals were consistent with the structure 7β -acetoxy- 6β -benzoyloxy- 3α -(3,4,5-trimethoxycinnamoyloxy)-tropane (**3d**) for this alkaloid.

Fraction 13 was obtained from residues of fraction 12; the constituent alkaloid was identified as 6β -benzoyloxy- 3α -(3,4,5-trimethoxycinnamoyloxy)tropane (**2d**), a known alkaloid of *E. monogynum* root-bark [5].

Of the roots of the *Erythroxylum* spp. so far studied in detail, those of *E. zambesiacum* have yielded the largest number of alkaloids. In general the alkaloid mixture resembles that of *E. monogynum* [5], but the range is extended by the presence of nortropans and the involve-

ment of acetic and phenylacetic acids. It was previously suggested [4] that the esterifying acids might provide a basis for chemotaxonomic groupings within this genus of some 120 species, a large proportion of which probably contain tropane esters (28 species examined in these laboratories have, without exception, contained such bases). The acid components of the alkaloids of the roots of species so far examined are given in the Table 1. The examples represent seven of Schulz's sections of the genus and cover the principal geographic areas of distribution. Distinctive patterns of acids are particularly evident for the four sections originating from the S.E. Africa area of diversification of the genus. Further studies are required to validate such chemotaxonomic implications. Table 1 relates only to the root-barks; other acids also occur as components of the alkaloids of the aerial parts of particular species [7, unpublished results]. The isolation of **2c** is of interest as it appears to be the first reported tropane alkaloid common to *Erythroxylum* and the Proteaceae.

EXPERIMENTAL

Instrumentation and the chemical methods for the synthesis and hydrolysis of esters were as recorded in ref. [4]. The prep. TLC systems employed all involved 0.5 mm layers and were A, Al_2O_3 with $\text{Et}_2\text{O}-\text{EtOH}$ (1:1); B, silica gel with $\text{CHCl}_3-\text{Et}_2\text{NH}$ (9:1); C, silica gel with $\text{Me}_2\text{CO}-\text{H}_2\text{O}-\text{concNH}_3$ (80:15:2).

Plant material. Collected and authenticated by Mr D. B. Fanshawe, Division of Forest Research, Kitewe, Zambia, 1972. Collection area, Katombosa Forest Reserve. Material supplied via the Tropical Development and Research Institute, London.

Extraction and isolation of alkaloids. In typical procedures powdered root-bark (100 g) was mixed with $\text{Ca}(\text{OH})_2$ (20 g) and H_2O (40 ml), allowed to stand for some hrs, and exhaustively extd with Et_2O . The solvent was removed and the residue redissolved in a minimum vol. of Et_2O and submitted to chromatography [kieselguhr (15 g), 0.5 M Pi buffer soln, pH 6.8. (7.5 ml)] with petrol (bp 40–60°), Et_2O and CHCl_3 as successive solvents. Further resolution of the alkaloid mixt. giving in all 13 fractions, was achieved by prep. TLC (System B and C above). Fractions 1–8 were principally associated with the CHCl_3 eluate above, fraction 9 with the later Et_2O eluate, fraction 10 with the main Et_2O eluate and fractions 11–13 with the petrol eluate.

3α -(3,4,5-Trimethoxybenzoyloxy)nortropane (**1a**). Fraction 1 contained one major alkaloid, R_f 0.25 (System C), and was purified as its picrate (serrated plates from $\text{EtOH}-\text{H}_2\text{O}$), mp 235° decomp. (softens 225°); EIMS (probe) 70 eV, m/z (rel. int.): 321.1595 $[\text{M}]^+$ ($\text{C}_{17}\text{H}_{23}\text{NO}_5$ requires M_r 321.1576 (1), 212.0692 $[\text{TmbA}]^+$ (calc. for $\text{C}_{10}\text{H}_{12}\text{O}_5$: 212.0685) (100), 197.0456 $[\text{TmbA}-\text{Me}]^+$ (calc. for $\text{C}_9\text{H}_9\text{O}_5$: 197.0450) (24), 195.0652 $[\text{Tmb}]^+$ (calc. for $\text{C}_{10}\text{H}_{11}\text{O}_4$: 195.0657) (8), 110.1018 (calc. for $\text{C}_7\text{H}_{12}\text{N}$: 110.0970) (44), 82, 80. The R_f value (System C) and the IR spectrum of the base were identical to those of the authentic semi-synthetic base (see below); mmp of the two picrates 235° decomp.

Preparation of synthetic 1a picrate. An authentic sample of **1a** prepared by demethylation of the corresponding N-methyl compound gave a picrate (plates from $\text{EtOH}-\text{H}_2\text{O}$), mp 244° decomp. (Found: C, 50.0; H, 4.9; N, 10.5. $\text{C}_{17}\text{H}_{23}\text{NO}_5 \cdot \text{C}_6\text{H}_3\text{N}_3\text{O}_7$ requires C, 50.2; H, 4.8; N, 10.2%).

Fractions 2–4. The base contained in fraction 2 was not identified; that contained in fraction 3 had R_f 0.39 (System C) and was characterized as 3α -(3,4,5-trimethoxybenzoyloxy)tropane (**1b**) by comparison of its R_f values, mp of picrate, high resolution MS data and IR spectrum with those of the authentic compound.

Table 1. Acids recorded as ester-components of tropanols in *Erythroxylum* root-barks

	<i>E. zambesiacum</i> [this paper]	<i>E. dekindtii</i> [13]	<i>E. hypericifolium</i> [4]	<i>E. macrocarpum</i> [10]
Section (Schulz)	Melanocladus	Lagynocarpus	Venelia	Pachylobus
Source	Zambesi valley TmbA (5) Benzoic (5) TmcA (4) Phenylacetic (1) Acetic (1)	Angola Isovaleric (2) α -Methyl- butyric (1) Phenylacetic (1) Furoic (1)	Mauritius Phenylacetic (5) 3-Hydroxy- phenylacetic (1) Acetic (1)	Mauritius Benzoic (1)
	<i>E. sideroxyloides</i> [10]	<i>E. monogynum</i> [5]	<i>E. cumanense</i> [11]	<i>E. coca</i>
Pachylobus Mauritius	Sethia S. India	Archerythroxylum Venezuala	Archerythroxylum Peru, etc	Coelocarpus N. Queensland Australia
Benzoic (1) Butyric (1)	TmbA (2) TmcA (2)	Benzoic (2) Phenylacetic (2)	No esters reported in root-bark	Benzoic (1) [7] Cinnamic(1) [unpublished]
	Benzoic (1)	TmbA (1)		
		Acetic (1)		

Figure in parenthesis indicates the number of alkaloids involving this acid. TmbA = 3,4,5-trimethoxybenzoic acid, TmcA = 3,4,5-trimethoxycinnamic acid.

Fraction 4, associated with fraction 3, had R_f 0.35 (System C) and was identified as 3 α -(3,4,5-trimethoxycinnamoyloxy)tropane by comparison of R_f values and the CIMS with those of the authentic compound.

(+)-3 α -(3,4,5-Trimethoxybenzoyloxy)tropan-6 β -ol (**2a**). The principal base of fraction 5 had R_f 0.41 (System C) and was purified by recrystallization of the picrate (rods from EtOH-H₂O), mp 214° (Found: C, 49.7; H, 5.1; N, 9.8. C₁₈H₂₅NO₆·C₆H₃N₃O₇ requires C, 49.6; H, 4.9; N, 9.7%). For the free base [α]_D²⁰ + 3.5° (EtOH; c1.5); IR $\nu_{\text{max}}^{\text{KBr}}$ cm⁻¹: 3500 (OH), 1710 (ester C=O); EIMS (probe) 70 eV, m/z (rel. int.): 351.1678 [M]⁺ (C₁₈H₂₅NO₆ requires M_r 351.1682) (1), 307.1411 [4a]⁺ (calc. for C₁₆H₂₁NO₅: 307.1420) (1), 212.0677 [TmbA]⁺ (5), 195.0653 [Tmb]⁺ (8), 140.1072 (calc. for C₈H₁₄NO: 140.1075) (35), 122, 110, 94 (100); ¹H NMR (250 MHz, CDCl₃): δ 1.61, 1.67, 1.75 and 1.81 (4H, 4 \times dd, H₂-2, H₂-4), 2.29 (2H, *m*, H₂-7), 2.56 (1H, *s*, exchangeable with D₂O, OH-6), 2.63 (3H, *s*, NMe), 3.16 (1H, *br s*, H-1), 3.40 (1H, *br s*, H-5), 3.92 (9H, *s*, 3 \times OMe), 4.67 (1H, *dd*, H-6), 5.23 (1H, *t*, J = 5.0 Hz, H-3), 7.28 (2H, *s*, *o*-Tmc-H₂). Alkaline hydrolysis of the base gave tropane-3 α ,6 β -diol (TLC) and 3,4,5-trimethoxybenzoic acid (mp, mmp); EIMS, m/z 212.0672 [M]⁺, (calc. for C₁₀H₁₂O₅; M_r 212.0685). The 6-phenylacetate derivative of the alkaloid, purified by TLC, gave MS m/z 469.2077 [M]⁺ (C₂₆H₃₁NO₇ requires M_r 469.2101), 307.1370 [4a]⁺.

(-)-3 α -Phenylacetoxytropan-6 β -ol (**2b**). Fraction 6 was deposited as a semi-solid from the picrate mother-liquors of **2a**; the liberated base gave [α]_D²⁰ -13.5° (EtOH; c1.1) and had the spectroscopic properties (MS, NMR) of authentic **2b**. The 6 β -phenylacetate of natural **2b** prepared by standard means gave a picrate (flat needles/from EtOH-H₂O, 1:1), mp 134° (Found: C,

57.6; H, 4.8; N, 8.8. Calc. for C₂₄H₂₇NO₄·C₆H₃N₃O₇·C, 57.9; H, 4.9; N, 9.0); mmp with picrate (mp 130°) of diester prepared from (\pm)-diol 130–135°; mmp with picrate (mp 151°) of (-)-diester prepared from the (+)-diol 135–146°.

3 α -(3,4,5-Trimethoxybenzoyloxy)tropan-6 β ,7 β -diol (**3a**). Fraction 7 contained **3a** as a principal component of the alkaloid mixture of the bark. The properties of the base (IR, MS, NMR, picrate characters and hydrolytic products) were as recorded [5] for the characterization of **3a** from *E. monogynum*.

6 β -Benzoyloxytropan-3 α ,7 β -diol (**3b**). The main component of fraction 8 gave a broad band R_f 0.62 by prep. TLC, System A. It was further purified by conversion to the picrate, mp 213° (plates from EtOH-H₂O) (Found: C, 49.3; H, 4.6; N, 10.9. C₁₅H₁₉NO₄·C₆H₃N₃O₇ requires: C, 49.8; H, 4.4, N, 11.1%). IR $\nu_{\text{max}}^{\text{KBr}}$ cm⁻¹: 3495 (OH), 1725 (ester C=O); EIMS (probe) 70 eV, m/z (rel. int.): 277.1315 [M]⁺ (C₁₅H₁₉NO₄ requires M_r 277.1314) (1), 229 (picric acid), 154.0839 (calc. for C₈H₁₂NO₂: 154.0868), 138.0909 (calc. for C₈H₁₂NO: 138.0919) (100), 113.0814 [4b]⁺ (calc. for C₆H₁₁NO: 113.0841) (48), 105.0319 (calc. for C₇H₅O: 105.0340) (56), 96 (49), 95, 94; ¹H NMR (250 MHz, CDCl₃), base recovered from picrate, δ 2.18, and 2.24 (4H, *dd*, H₂-2, H₂-4), 2.59 (3H, *s*, NMe), 3.16 and 3.32 (2H, 2 \times *br s*, H-1, H-5), 4.12 (1H, *t*, J = 4.8 Hz, H-3 α), 4.87 (1H, *d*, $J_{6\alpha,7\alpha}$ 6.2 Hz, 7 α -H), 5.82 (1H, *d*, $J_{6\alpha,7\alpha}$ 6.2 Hz, 6 α -H), 7.5 (3H, *m* Bz-H₃), 8.1 (2H, *m*, Bz-H₂).

(+)-6 β -Benzoyloxytropan-3 α -ol (**2c**). Five components were sep'd by prep. TLC from fraction 9; the principal base had R_f 0.26, and 0.55 (systems B and C, respectively), was dextrorotatory and readily formed a picrate, mp 180° (needles from EtOH). The chromatographic and spectroscopic (IR, MS) properties were identical with those of a sample of semi-synthetic (\pm)-6 β -

benzoyloxytropan-3 α -ol prepared in other studies [12].

3 α -Trimethoxybenzoyloxynortropan-6 β -ol (2f) was isolated from fraction 10. [R_f 0.24 (system B)] and afforded a picrate, mp 201° (prisms from EtOH-H₂O) (Found: C, 48.7; H, 5.0; N, 9.8. C₁₇H₂₃NO₆·C₆H₃N₃O₇ requires: C, 48.8; H, 4.6; N, 9.9%). IR $\nu_{\text{max}}^{\text{KBr}}$ cm⁻¹: 3520 (OH), 1716 (ester C=O); EIMS (probe) 70 eV, m/z (rel. int.): 337.1512 [M]⁺ (C₁₇H₂₃NO₆ requires M_r 337.1525) (2), 229 (picric acid) (16), 212.0685 [TmbA]⁺ (56), 195.0660 [Tmb]⁺ (47), 142.0873 (calc. for C₇H₁₂NO₂, 142.0868), 125.0842 (calc. for C₇H₁₁NO: 125.0840) (100), 108 (42), 80 (12). For the regenerated base, ¹H NMR (250 MHz, CDCl₃) δ 1.50–1.78 (6H, *m*, H₂-2, H₂-4, H₂-7), 2.50 (2H, *exch.* with D₂O, NH, OH) 3.47 (1H, *br s*, H-1), 3.68, (*d*, J = 6.3 Hz, H-5), 3.91 (9H, *s*, 3 \times OMe), 4.56 (1H, *dd*, J = 6.9 and 2.2 Hz, H-6), 5.00 (1H, *m*, H-3), 7.28 (2H, *s*, *o*-Tmb-H₂).

6 β -Benzoyloxy-3 α -(3,4,5-trimethoxycinnamoyloxy)tropan-7 β -ol (3c). A base of fraction 11 of the eluate from the initial column had R_f 0.70 (system B) and R_f 0.83 (system C); it was purified by repeated recrystallization of the picrate, mp 234° from EtOH-H₂O. IR $\nu_{\text{max}}^{\text{KBr}}$ cm⁻¹: 3460 (OH), 1725, 1715 (2 \times ester C=O); EIMS (probe) 70 eV, m/z (rel. int.): 497 [M]⁺ (2), 333 (2), 238.0843 [TmcA]⁺ (calc. for C₁₂H₁₄O₅: 238.0842) (12), 221.0797 [Tmc]⁺ (calc. for C₁₂H₁₃O₄: 221.0814) (8), 138 (42), 122.0968 (calc. for C₈H₁₂N: 122.0969) (9), 122.0365 (calc. for C₇H₆O₂: 122.0368) (5), 105 (74), 94 (100). ¹H NMR (250 MHz, CDCl₃), base recovered from picrate, δ 2.24 (4H, *m*, H₂-2, H₂-4), 2.75 (3H, *s*, NMe), 3.2 and 3.4 (2H, 2 \times *br s*, H-1, H-5), 3.90 and 3.95 (9H, 2 \times *s*, 3 \times OMe), 4.87 (H, *br s*, H-7), 5.25 (1H, *t*, H-3), 5.87 (H, *br s*, H-6), 6.35 and 7.77 (2H, 2 \times *d*, J = 16 Hz, CH=CH), 6.9 (2H, *s*, *o*-Tmc-H₂), 7.50 (3H, *m*, Bz-H₃), 8.06 (2H, *m*, Bz-H₂).

7 β -Acetoxy-6 β -benzoyloxy-3 α -(3,4,5-trimethoxycinnamoyloxy)tropane (3d). Fraction 12 was sepd from fraction 11 by prep. TLC and had R_f 0.76 (System B) and 0.87 (System C). The principal base was purified by picrate formation, [plates, mp 155° from EtOH-H₂O] (Found: C, 54.2; H, 4.9; N, 7.2. C₂₉H₃₃NO₉·C₆H₃N₃O₇ requires: C, 54.7; H, 4.9; N, 7.3%). IR $\nu_{\text{max}}^{\text{KBr}}$ cm⁻¹: 1750, 1725, 1715 (3 \times ester C=O); EIMS (probe) 70 eV, m/z (rel. int.): 337.1512 [M]⁺ (C₁₇H₂₃NO₆ requires M_r 337.1525) (5), 434 [M - PhCO]⁺ (1), 333.1599 [4d]⁺ (calc. for C₁₈H₂₃NO₅: 333.1576) (2), 318 (13), 302.1390 [M - (MeO)₃C₆H₂CH:CHCOO]⁺ (13), 238.0834 [Tmc]⁺ (calc. for C₁₂H₁₄O₅: 238.0842) (6), 229 (picric acid), 181 (1), 180.1018 [MeCOO·C₈H₁₁N]⁺ (calc. for C₁₀H₁₄NO₂: 180.1025) (14), 138 [C₈H₁₂NO]⁺ (14), 122.0963 [C₈H₁₂N]⁺ (10), 122.0356 [C₇H₆O₂]⁺ (5), 105.0293 [C₇H₅O]⁺ (54) 94 (100), 43 (3). Base [α]_D²⁰ -24.0° (EtOH: Me₂CO (1:1), c 0.5) ¹H NMR (250 MHz, base in CDCl₃), δ 1.87 (3H, *s*, COMe), 2.23 (4H, *m*, H₂-2, H₂-4), 2.60 (3H, *s*, NMe), 3.27 and 3.38 (2H, 2 \times *br s*, H-1, H-5), 3.81 and

3.85 (9H, 2 \times *s*, 3 \times OMe), 5.21 (1H, *t*, J = 4.5 Hz, H-3), 5.77 and 5.92 (2H, 2 \times *d*, J = 6.4 and 6.2 Hz, respectively, H-6 α , H-7 α), 6.29 and 7.74 (2H, 2 \times *d*, J = 16.1 Hz, CH=CH), 6.88 (2H, *s*, *o*-Tmc-H₂), 7.45 (3H, *m*, Bz-H₃), 7.95 (2H, *m*, Bz-H₂).

6 β -Benzoyloxy-3 α -(3,4,5-trimethoxycinnamoyloxy)tropane (2d). A minor component of fraction 12 was partially purified by repeated recrystallization of the picrate, mp 234° from same R_f values (3 systems) and the same mass spectroscopic characteristics as authentic 6 β -benzoyloxy-3 α -(3,4,5-trimethoxycinnamoyloxy)tropane isolated from natural sources.

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